Abstracts of forthcoming articles

Crystal Structure and Fast Ionic Conduction of TlZrF₅. D. AVIGNANT, I. MANSOURI, R. CHEVA-LIER, AND J. C. COUSSEINS,* Groupe de Cristallographie et de Chimie des Solides, Universite de Clermont-Ferrand II, B.P. 45, 63170 Aubiere, France. TlZrF₅ crystallizes in the monoclinic system with unit-cell dimensions a = 8.112(1) Å, b = 7.927(3) Å, c = 7.929(1) Å, $\beta = 123.99(1)^\circ$, and space group $P2_1/c$ (No. 14); Z = 4. The structure was solved by conventional Patterson and Fourier methods and refined by full-matrix least squares to a conventional R of 0.057 ($R_{\omega} = 0.063$). The structure consists of sheets (ZrF₅)⁻ that may be described as edge-shared and corner-shared bicapped trigonal prism (ZrF₈). The sheets run parallel to the $y\partial z$ plane and are bonded together by the Tl ions which are surrounded by 12 F⁻ atoms. The ionic conductivity of TlZrF₅ and TlHfF₅ has been investigated by complex impedance measurements and the relationships between structure and fast ionic conduction are discussed.

Oxidation in the γ Phase of Spinels Containing Iron. II. Influence of Defects on the Oxidation Kinetics and Electrical Properties. B. GILLOT,* F. CHASSAGNEUX, AND A. ROUSSET, Laboratoire de Recherches sur la Reactivite des Solides, Faculté des Sciences Mirande, B.P. 138, 21004 Dijon Cédex, France. After a review of the distribution of vacancies in defect phases resulting from γ -Fe₂O₃, the authors give several examples drawn from oxidation kinetics and electrical properties where the vacancies play a basic part due to their concentration as well as their location. The decrease in chemical diffusion coefficient with increase in vacancy content and the variation of the exponent from the pressure law with the extent of association are dependent on concentration while the nature of the electron hopping between Fe²⁺ and Fe³⁺ ions is governed by the location of vacancies in the two types of sites in the spinel lattice.

Isothermal Decomposition of Ternary Oxide $A_x B_y O_z$ on Isobar Stability of Perovskite ABO₃ (A = La, Sm, Dy; B = Mn, Fe) in Reducing Atmosphere. T. NAKAMURA, Research Laboratory of Engineering Materials, Tokyo Institute of Technology, 4259 Nagatsuta-cho, Yokohama 227, Japan. The isothermal decomposition of any ternary oxide $A_x B_y O_z$ on liberation of n moles of oxygen at a constant pressure is found to be driven by the mixing entropy $\Delta S_m = -nR \ln P_{O_2}$ of the total entropy change $\Delta S = \Delta S^\circ + \Delta S_m$. The stability of $A_x B_y O_z$ toward isothermal decomposition into a biphasic solid mixture is derived from the equilibrium condition $\Delta G^* = 0$ as functions of standard changes ΔH° and ΔS° . Assuming $\Delta S^\circ = 44n$ and calculating ΔH° in terms of lattice energies $U(ABO_3)$ and $U(A_2O_3)$, the stability of perovskites $St(ABO_3) = -\log P_{O_1^*} (A = La, Sm, Dy; B = Mn, Fe)$ is given as a function of the ionic radius of the A^{3+} ion. The calculated stability agrees with the observed. The effect of electronic entropy change ΔS_e on ΔS° is demonstrated for AFeO₃ (A = La, Sm, Dy).

Neutron Powder Diffraction on α - $\Pi_4 CrI_6$ and β - $\Pi_4 CrI_6$. H. W. ZANDBERGEN, Gorlaeus Laboratories, University of Leiden, P.O. Box 9502, 2300 RA Leiden, The Netherlands. α - $\Pi_4 CrI_6$ (a = 9.132(1), c = 9.667(1) Å, Z = 2, P4/mnc at 293 K) adopts a distorted Π_4 HgBr₆ structure. In α - $\Pi_4 CrI_6$ a random distribution of Jahn Teller distorted octahedra occurs, which are elongated perpendicular to the c axis. Between 77 and 4.2 K a phase transition occurs. In β - $\Pi_4 CrI_6$ (a = 12.941(3) Å, b = 12.596(3) Å, c = 9.602(2) Å, Z = 4, Cccm at 4.2 K) the directions of elongation of the octahedra are ordered. The structure is very much related to that of α - $\Pi_4 CrI_6$. A three-dimensional magnetic ordering takes place at 2.7(2) K. The magnetic space group at 1.2 K is $C_122'2'$. The magnetic sublattices are present forming two independent magnetic lattices which have no interaction due to the antiparallel ordering.

Note: Asterisks indicate author to be addressed.